Pulsed-Laser Deposited 35 Bi(Mg_{1/2}Ti_{1/2}) O₃-65 PbTiO₃ Thin Films—Part I: Influence of Processing on Composition, Microstructure, and Ferroelectric Hysteresis

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Abstract-35 Bi(Mg_{1/2}Ti_{1/2})O₃ - 65 PbTiO₃ (35 BiMT-65 PT) is a potential candidate material for a high-temperature nonvolatile ferroelectric memory. For pulsed-laser deposited 35 BiMT-65 PT films with the perovskite structure, it was found that as the chamber pressure during deposition decreased, the Mg and Pb contents in as-deposited films drop, while the concentration of Bi increases. Concurrently with the change in composition, the remanent polarization P_r increases 64% to $\approx 21 \ \mu \text{C/cm}^2$ and the polarization-electric field loops rotated counterclockwise as the deposition pressure increases. Decreasing the seed layer thickness from 36 to 16 nm led to a decrease in P_r to $\approx 14 \ \mu$ C/cm². Adjusting the target composition allowed the deposition of films which had near-stoichiometric Bi and Mg concentrations, but in all cases, the grown films were lead deficient. These films had remanent polarizations of 18–20 μ C/cm². If the lead content of the target was increased too far, the remanent polarization decreased, possibly due to the need to evolve more PbO from defective growth layers. Finally, the deposition rate showed no substantial effect on the film composition, but did have a significant impact on the ferroelectric properties. As the deposition rate decreased, the P_r increased to $\approx 22 \ \mu$ C/cm² due to enhanced crystalline quality. At laser frequencies of 5 Hz, a Mg-rich pyrochlore phase begins to form and films showed a maximum $P_r \approx 22 \ \mu \text{C/cm}^2$. The processingcomposition behavior is explained via preferential adsorption of Bi on the A-site, which results in lead vacancies.

Index Terms—A-site deficient perovskite, bismuth magnesium titanate–lead titanate, high-temperature ferroelectric random access memory (FeRAM).

I. INTRODUCTION

S INCE the 1990s, lead zirconate titanate (PZT) has provided a stable platform for the production of nonvolatile ferroelectric random access memory (FeRAM) technology, but new materials are being sought to replace it [1]. Currently, Texas Instruments produces state-of-the-art embedded PZT FeRAM on 130-nm node CMOS, with an operating life of

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10 years and 1000 h at 85 °C and 125 °C, respectively [2]. Nonvolatile ferroelectric memories are also available from Cypress, Fujitsu, Rohm, and Matsushita (Panasonic) [3]–[6]. Increasing the operating temperature of FeRAM would extend its utility to harsher environments, including automotive applications. While a significant amount of research is dedicated to the scaling of thin film ferroelectrics for FeRAM [2], [7]–[12], high-temperature functionality of the films has not been explored as extensively. Ultimately, the practical limiting use temperature is a fraction of the Curie temperature (T_c). Films with high Curie temperatures and large switchable polarizations are of potential interest for high-temperature FeRAM.

Eitel *et al.* [13], [14] and Randall *et al.* [15] predicted, based on the Goldschmidt tolerance factor [16], that several Bi-based-PbTiO₃ solid solutions should have higher transition temperatures than PZT at the morphotropic phase boundary (MPB). This was confirmed experimentally first for BiScO₃-PbTiO₃, which acts as a PZT-analog, but with an MPB phase transition temperature ~100 °C higher than PZT [14], [17]. Subsequently, multiple perovskite systems have been demonstrated to exhibit a T_c higher than that of PbTiO₃ over part of the solid solution range [18]–[23].

Stringer *et al.* [21] and Stringer [24] proposed three categories of Bi-based-PbTiO₃ solid solutions. Category 1 corresponds to systems similar to PZT, where the maximum transition temperature exists at pure PbTiO₃. The third category corresponds to systems similar to BiFeO₃ where the maximum in T_c increases with decreasing PbTiO₃ concentration due to an enhancement in tetragonality. Finally, the second category has a maximum in T_c between the end members, the origins of which are not yet understood.

The focus in this paper is on a category 2 solid solution system: x Bi(Mg_{1/2}Ti_{1/2})O₃ - (1-x)PbTiO₃ (BiMT-PT). Fig. 1 shows the phase diagram for the BiMT-PT system. The bulk of the literature has focused on compositions around the MPB. Suchomel and Davies [18] utilized the tolerance factor of the non-PbTiO₃ end member of known systems to narrow the predicted region for the MPB location (close to x = 62-64 in BiMT-PT). Randall *et al.* [15] refined the MPB compositional region utilizing piezoelectric (d_{33}), poled permittivity, and X-ray diffraction (XRD) measurements. They found

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Fig. 1. Phase diagram of the BiMT-PT system. Reported data (blue upward triangle [21], black square [15], purple diamond [18], and green downward triangle [25]) are based on combined high-temperature XRD, differential scanning calorimetry, and dielectric permittivity versus temperature measurements. The blue line corresponds to a second-order polynomial fit as performed by Stringer *et al.* [21], but for all reported data. The shaded region corresponds to third-order polynomial fits and the uncertainty in the transition temperature. See supplemental materials for further details.

that it was hard to pinpoint the exact MPB composition due to the propensity for formation of grains with core-shell structures associated with compositional gradients [15]. BiMT-PT ceramics were reported to have remanent polarizations of $\approx 33-38 \ \mu\text{C/cm}^2$ (at the MPB of x = 0.36 PT) [15], [26] and 25 μ C/cm² (at x = 0.5 PT) [27]. Compositions around the MPB had increased P_r and decreased coercive field E_c and were resistant to thermal depoling to at least 175 °C [26], [28]. Some reports suggest that there is limited domain wall mobility for some BMT-PT with high transition temperatures due to large lattice distortions [26] or defects. In short, BiMT-PT materials demonstrate promise as an alternative high-temperature ferroelectric without requiring use of scandium.

The literature on thin film Bi-based-PbTiO₃ solid solutions is considerably smaller than that on bulk ceramics. Zhong et al. [42] grew 600-nm 100c oriented 63 BiMT-37PT thin films with a P_r of 24 μ C/cm² and a relative permittivity ε_r maximum at \approx 415 °C. Zhang *et al.* [28] measured undoped and 0.5-mol% Mn doped 20 BiMT - 80 PT thin films, reporting a stable polarization-electric field response to 125 °C and fatigue resistance up to 10^8 cycles. Oikawa *et al.* [29] and Tanaka *et al.* [30] studied 111_c epitaxial BiMT films, a composition where the perovskite phase is metastable under ambient conditions. Films of Oikawa et al. [29] demonstrated little change in P_r over the thickness range of 50–800 nm [30]. Other Bi(Me' Me'')O₃ - PbTiO₃ solid solution thin films have also been produced, including BiScO₃-PbTiO₃ (BiS-PT) [31], BiInO₃ (BiI-PT) [32], [33], BiAlO₃-PbTiO₃ (BiA-PT) [34], BiFeO₃-PbTiO₃ (BiF-PT) [35]–[37], and Bi(Zn_{1/2}Zr_{1/2})O₃-PbTiO₃ (BiZZ-PT) [38]. Of the studied Bi(Me'Me'')O₃-PbTiO₃ thin films, BiS-PT, BiMT-PT, and BiI-PT are materials which may be able to withstand harsh environments for actuators, sensors, and FeRAM devices. 0.5-mol% Mndoped x BiI-(1-x)PT (x = 0.1-0.25) exhibit high transition temperatures (558 °C-633 °C) and P_r (25-44 μ C/cm²), but a large frequency dispersion in tan(δ) with temperature from 1 kHz to 1 MHz above 200 °C. 001_c epitaxial 60 BiScO₃ -40 PbTiO₃ demonstrate a $P_r \approx 42 \ \mu$ C/cm², $\varepsilon_{r,max}$ at $T \approx 455$ °C, and maintains a low loss (<0.1 at 100 kHz) up to 550 °C [31]. BiA-PT [34], BiZZ-PT [38], and BiMT-PT [39] all demonstrate leakage currents of 10⁻⁵ A/cm² or less at 150 kV/cm, at least an order of magnitude less than that reported for 0.6 BiFeO₃ – 0.4 PbTiO₃ thin films [35], presumably due to the lower likelihood of electron hopping in the iron-free compositions.

The possibility for BiMT-PT ferroelectrics with large remanent polarizations, a stable high-temperature polarization, and a relatively thickness-independent P_r to film thicknesses of 50 nm are encouraging for developing next-generation high-temperature FeRAM. However, there is still a significant gap in knowledge in the development of BiMT-PT ferroelectrics. As shown in Fig. 1, there is uncertainty in T_c as a function of composition, particularly near $T_{c,\max}$. This is particularly notable for the thin film literature. For example, Liu and Zuo [40], Liu et al. [41] reported unseeded and seeded 63 BiMT - 37 PT with $P_r \approx 18 \ \mu \text{C/cm}^2$ and $\approx 14-15 \ \mu C/cm^2$, respectively, but Zhong *et al.* [42] report ≈ 8 and $\approx 24 \ \mu$ C/cm², respectively, for the same composition. Liu et al.'s films show significantly increased leakage compared to those in bulk ceramics [43]. In addition, there are discrepancies between reported film and crystal properties [31], [44]–[48].

This paper investigates a relatively unexplored composition in the BiMT-PT solid solution, with an emphasis on processing-structure-property relationships. In particular, one aspect that has little attention in the literature is the defect chemistry of systems in which both Pb and Bi occupy the A-site of the perovskite lattice, and the role of A-site point defects on electrical properties. Aliovalent point defects in Pb-based perovskites play a significant role in controlling domain wall mobility though V_{Pb}'' have been shown to only have a minor influence on the electrical response in PZT [49]. The 35 BiMT - 65 PT was chosen as a model system, as it is near $T_{c,max}$ in the BiMT-PT system. This paper focuses on the processing-composition-ferroelectric hysteresis response of these materials. A follow-on paper focuses on the thickness dependence of the electrical properties, hightemperature polarization-electric field (P-E) behavior, and charge injection of A-site deficient 35 BiMT - 65 PT thin films [50].

II. EXPERIMENTAL PROCEDURE

The 35 BiMT - 65 PT films were deposited on PbTiO₃/Pt/Ti/SiO₂/Si substrates. Either undoped or 5-mol% La-doped PbTiO₃ seed layers were deposited on the Pt-coated Si substrates using the sol–gel method to facilitate perovskite nucleation. Lead acetate trihydrate, titanium isopropox-ide (97%), lanthanum acetate hydrate, 2-methoxyethanol (2-MOE), and acetylacetone (ACAC) (all from Sigma-Aldrich Inc., St. Louis, MO, USA) were used as raw materials.

 TABLE I

 List of Target Compositions Investigated

Target Name	Pb Excess (mol%)	Bi Excess (mol%)	Mg Excess (mol%)
20Pb10Bi0Mg	20%	10%	0%
30Pb15Bi5Mg	30%	15%	5%
48Pb20Bi10Mg	48%	20%	10%
65Pb20Bi10Mg	65%	20%	10%
85Pb20Bi10Mg	85%	20%	10%

Lead acetate (40-mol% excess) and lanthanum acetate were dissolved in 2-MOE and distilled in a rotary evaporator under Ar gas flow until precipitation. A solution was made by adding titanium isopropoxide and 2-MOE to the precipitated powder with half the molarity of the final solution. After refluxing for 3 h, the solution was distilled, and ACAC (22.5 and 11.25 vol% for a 0.15- and 0.075-M solution, respectively) and 2-MOE was added to reach the final solution molarity. The solution was spun on the substrate at 3000 rpm for 45 s, pyrolyzed at 225 °C and 400 °C for 2 min each, and crystallized at 600 °C under oxygen flow for 1 min. The undoped 0.15-M and a doped 0.075-M solution produced film thicknesses of 36 and 16 nm, respectively.

Ceramic targets with varying compositions were batched using standard ball milling, calcination, and sintering. Briefly, Pb₃O₄, 4MgCO₃Mg(OH)₂.4H₂O (40%–43.5% MgO, Alfa Aesar), TiO₂ (cr-el grade, 99.9%, Ishihara Sangyo Kaisha), and Bi₂O₃ (varistor grade, 99.99%, MCP Metal Specialties) were batched in a stoichiometric ratio and balled milled in ethanol for 24 h. After drying at 85 °C, the powder was calcined at 775 °C for 4 h. Varying amounts of Bi₂O₃, PbO (Pb₃O₄ calcined at 700 °C for 1 h), and MgCO₃ excesses were added to the stoichiometric batch to compensate for losses during deposition. Table I gives the batched compositions of the targets employed. The powders were vibratory milled for 12 h, dried, pressed into pellets, and sintered at 825 °C for 2–4 h. Ignition losses were taken into account prior to batching.

Pulsed laser deposition (PLD) was used to deposit 35 BiMT - 65 PT films on the seeded substrates. For deposition, the chamber was pumped down to $\approx 10^{-6} - 10^{-7}$ Torr. A 90%/10% O₂/O₃ mixed gas was employed as the background atmosphere during deposition, with chamber pressures between 60 and 340 mTorr. A KrF excimer laser (Lambda Physik Compex 102, Fort Lauderdale, FL, USA) with an energy density of $\approx 1.6 \pm 0.1$ J/cm² was used. The deposition time was 10 min for samples deposited at 10 Hz at 340-mTorr deposition pressure. The laser spot size and laser measured energy are $\approx 11.3 \text{ mm}^2$ and 180 mJ. The deposition time was adjusted to grow films of approximately the same thickness for different processing variables. For samples deposited under different laser deposition frequencies, the number of laser pulses was kept constant. For all samples, the target to substrate distance was maintained at 6 cm and the substrate temperature was maintained at 700 °C. Substrates were bonded to the heater block using silver paste to insure good thermal contact. These parameters were chosen based on conditions for growing BiS-PT and PMN-PT, as well as further optimization [31], [51].

The films' phase purity was studied using a Philips Empyrean X-ray diffractometer with $CuK\alpha$ radiation. Film compositions were determined from focused ion beam milled cross sections using energy dispersive X-ray spectroscopy (EDS) on an FEI Talos transmission electron microscope (TEM) operated at 200 kV in scanning TEM (STEM) mode. High angular dark field (HAADF) images were collected; the resulting image primarily shows mass contrast approximately proportional to the square of the atomic number of the elements present. EDS data were collected using the Bruker Super-X quad X-ray detectors for 5 min with a beam current of approximately 0.25-nA [see supplemental materials (supplemental S.II) for the description of the composition normalization)] EDS quantification was performed using the Bruker Esprit software with the manufacture's supplied sensitivity factors for standardless Cliff-Lorimer quantification. The sensitivity factor database is calculated based on the accelerating voltage used to collect the data and geometry factors of the instrument.

Electrical properties were tested using lithographically defined 75- μ m circular Pt top electrodes. The electrodes were deposited in a Lesker CMS-18 sputter tool at an Ar pressure of 2.5 mTorr, and patterned by liftoff. All electrodes were annealed at 600 °C for 1 min in flowing O₂. The thin film ferroelectric properties were studied using a Radiant Precision Multiferroic tester. *P*–*E* data were taken at 1 kHz. All positive-up-negative-down (PUND) [10] data were acquired with a 1-ms pulse and 1-ms delays between pulses.

III. RESULTS AND DISCUSSION

In PLD, there are a wide variety of interrelated processing variables that control the plume–substrate interactions, and influence the resulting film structure and properties. To explore this processing space, the substrate temperature was held fixed at 700 °C, the target to substrate distance at 6 cm and the laser fluence at 1.6 ± 0.1 J/cm². The deposition pressure, target composition, and laser deposition rate were then varied systematically.

Fig. 2 shows XRD patterns as a function of chamber pressure during deposition, along with a representative TEM cross section showing the elemental map. For all of the processing conditions in this paper, unless stated otherwise, the films are $\{100\}_c$ oriented with a columnar grain structure (see supplemental materials). Additional XRD for other examined processing parameters can be found in the supplemental materials. Fig. 2(a) shows some changes in the shape of the h00 family peaks in the XRD patterns as the deposition pressure changes. Given that all of the films are deep in the tetragonal phase field of the BiMT-PT solid solution, these changes were attributed to differences in domain state or a small change in the defect chemistry that influences the lattice parameter. (See supplemental materials for additional information.) Film roughness decreases with lower deposition pressure due to an increase in bombardment energy as



Fig. 2. (a) XRD patterns of films deposited between 40-mTorr and 400-mTorr background pressure. Peaks marked with a star come from the substrate. The film thickness ranged from \approx 450 to 900 nm as the deposition pressure increased. Deposition time was 10 min for all samples. (b) and (c) STEM-HAADF image and STEM-HAADF image overlaid with color EDS elemental maps of elements of interest for a film grown at a chamber pressure of 340 mTorr. The thin layer between the Pt and Bi mapping corresponds to the PbTiO₃ seed layer.

suggested by the Thornton structure zone mode (see supplemental materials for additional information) [52]. EDS elemental mapping shows the films do not exhibit large spatial elemental segregation.

A. Deposition Pressure-Composition-Ferroelectric Property Relationships

The background chamber pressure controls the mean free path and interaction of ablated species with the background gas, determining the energy and chemical/physical state of species arriving at the substrate surface [53]. In general, high levels of bombardment can cause lattice expansion, imprint formation, and loss of volatile species [54]–[57].

Fig. 3(a) shows the average composition for films deposited between 60 and 340 mTorr. The target utilized had 20-mol% and 10-mol% excess Pb and Bi, respectively. As the deposition pressure decreased, the following phenomena were observed. First, the Mg concentration decreases with decreasing background pressure. The Mg loss is attributed to scattering out of the plume, as it is a light ion. Lichtenwalner [53] demonstrated that the introduction of an ambient gas changes the angular distribution of species ejected from the target. As the chamber pressure is increased, the angular distribution becomes confined. The plume expands at low pressures, and only the central portion of it directly deposits on the substrate. Second, the Pb content also decreases as the chamber pressure drops, presumably as the increased bombardment levels cause more resputtering from the growing film. Finally, the Bi concentration increases as the pressure decreases. This is contrary to what is expected based on its comparatively high propensity for resputtering [56], [57], suggesting that another mechanism must be involved. All films were determined to be



Fig. 3. (a) Concentration of species X, where X is Pb, Mg, or Bi as a function of chamber background pressure. The compositions are normalized to the Ti concentration. A stoichiometric composition would have all cations equal to one. (b) 1-kHz ferroelectric hysteresis response of 670–720-nm-thick films with a 36-nm PbTiO₃ seed layer deposited between 60 and 340 mTorr. (c) PUND switching (P*) and (d) nonswitching (P^{\wedge}) charge with a 1-ms pulse and 1 ms 0-V holds.

A-site deficient, with a decrease in the overall A-site deficiency as the Bi-concentration increases.

The ferroelectric hysteresis loops shown in Fig. 3(b) for films between 670 and 720 nm in thickness indicate that P_r and maximum polarization (P_{max}) correlate with improved film stoichiometry [Fig. 3(a)]. P_{max} , P_r , and leakage current (see Part II [50]) increase with an increase in background deposition pressure. The increased P_{max} and P_r correlate with a counterclockwise rotation of the P-E loop, resulting in a P_r increase of 64% to 21.3 and $-22 \ \mu C/cm^2$. The correlation between deposition pressure and ferroelectric P-E response may be related to the excess Bi present at lower pressures. Bahri *et al.* [58] and Skanavi *et al.* [59] demonstrated that Bi doping lowers the Curie temperature in BaTiO₃. This, in turn, decreases the remanent polarization, as demonstrated by Ogihara *et al.* [60]. The polarization data suggest that the [Bi] strongly influences the P-E response in these films.

PUND measurements were used to quantify total switching (P^*) and nonswitching (P^\wedge) charges, respectively, as shown in Fig. 3(c) and (d). With increasing deposition pressure, at 500 kV/cm, the total P^* increases by $\approx 32 \ \mu$ C/cm²; however, P^\wedge increases by $\approx 11-14 \ \mu$ C/cm². While other sources may contribute to an increase in P^\wedge , the likely source is leakage, given to the relatively small differences in composition and structure from sample to sample. The increase in leakage is expected to be due to A-site vacancy-electron trapping and will be discussed in more detail in a subsequent paper [50].

B. Target Composition-Ferroelectric Property Relationship

A series of targets were made to compensate for the stoichiometric deficiencies in the grown films. Table I lists the batched composition for these targets. If either volatility or resputtering of PbO or Bi₂O₃ is the primary source for the A-site deficiency, then it should be possible to compensate for the nonstoichiometry via the target. A series of films were deposited using each target. All films were deposited at 340 mTorr, a 6-cm target to substrate distance, and a laser deposition rate of 10 Hz. These films were deposited on a 16-nm 5-mol% La-doped PbTiO₃ to reduce the influence of the seed layers on the resulting electrical properties. Fig. 4 shows the composition profile and ferroelectric response of the resulting films. Reducing the seed layer thickness resulted in a decrease in P_r to14.4 and $-15.2 \ \mu C/cm^2$ using a 20Pb10Bi0Mg target.

Fig. 4(a) shows that compensation of cation deficiencies is possible for most of the cations. The Bi and Mg compositions approach the ideal concentrations with respect to the Ti concentration on adding 20-mol% and 10-mol% excess, respectively. However, the Pb concentration shows only minor increases regardless of the amount of excess PbO added to the target. It is hypothesized that this may be due to competition between Pb and Bi adsorption onto the A-site.

Fig. 4(b) and (c) shows the ferroelectric response of films deposited from all targets. The control samples deposited from the 20Pb10Bi0Mg target shows the lowest P_r based on P-E (1 kHz) and PUND measurements. As the cation concentration ratios approach ideal, the P_r increases. However, excessive amounts of PbO in the target lead to a decrease in P_r . It is



Fig. 4. (a) Normalized composition of 600–700-nm films using a 16-nm (film deposited using a 20 Pb 10 Bi 0Mg target had a 36-nm PbTiO₃ seed for composition analysis) PbTiO₃ seed layer. The seed layer was avoided during the cross-sectional analysis. (b) Ferroelectric hysteresis (1 kHz) response as a function of target composition (all on 16-nm seed). (c) Sample PUND data for \approx 600–700-nm-thick films measured using 1-mS pulses and 1-mS 0-V holds.

believed that this is a consequence of a greater amount of disorder on the surface associated with the need to evolve more PbO from the surface during growth. The growing layers become more defective and the processing window shifts away from 10 Hz (see Section III-C). Also, the trend in P_r as a function of target composition was not identical to the



Fig. 5. (a) Composition of films deposited between at laser pulse frequencies of 5–20 Hz. (b) Ferroelectric P-E hysteresis (1 kHz) of films deposited between 1 and 20 Hz.

trend in crystallographic orientation as a function of pressure (see supplementary materials for additional details). The mechanism controlling this target composition dependence of P_r is unknown.

C. Laser Deposition Rate-Film Composition-Ferroelectric Property Relationships

The laser repetition rate also influences the film composition. Maria *et al.* [51] and Maria [61] demonstrated for PMN-PT films that lower laser repetition frequencies provide longer times for evolution of the most volatile species. Changing the laser repetition rate leads to a change in the growth rate as well, which can also change the crystalline quality of the resulting films. In particular, slower growth rates are expected to provide more time for atomic rearrangement, leading to improved ferroelectric properties.

Fig. 5 shows film compositions and resulting ferroelectric properties for growth from a 48Pb20Bi10Mg target. It is evident that changing the laser frequency between 5 and 20 Hz does not yield a significant difference in the resulting film compositions. However, there is a significant change in the film properties. The maximum P_r and P_{max} occur at 3 and 5 Hz ($\approx 22 \ \mu \text{C/cm}^2$), while there is a continuous decrease in P_r and P_{max} as the laser frequency increases up to 20 Hz.



Fig. 6. (a) HAADF-STEM image of cross section from sample from 35 BiMT-65 PT deposited at 5 Hz using a 48Pb20Bi10Mg target at 340 mTorr. Dark regions correspond to a Pb/Bi deficient secondary phase. (b) Mg quantitative EDS elemental map (atomic percentage). (c) Pb quantitative elemental EDS map (atomic percentage). Yellow circles highlight regions with decreased Pb concentrations. (d) Combined Pb and Mg elemental maps showing the overlap between Pb deficient regions and Mg rich regions.

The loop is noticeably lossier at 3 and 1 Hz. Similar results were found when utilizing targets with greater PbO excess compensation.

Films deposited at frequencies of ≤ 5 Hz contain a Pb deficient and Mg rich second phase (see Fig. 6). Unfortunately, XRD analysis was not sensitive enough to detect secondary phase peaks even in the 1-Hz sample, but the Pb deficiency suggests this is a pyrochlore phase. Interestingly, P_r and P_{max} remain high for the films grown at 5 Hz, regardless of the presence a minor amount of second phase. A similar phenomenon is reported in systems such as PMN-PT [51].

IV. DISCUSSION

The composition of films grown with volatile species can be either flux controlled or adsorption controlled. In fluxcontrolled growth, the film composition is dependent on the species provided to the growing film. In adsorption-controlled growth, the film composition is dependent on the adsorption kinetics, independent of the species flux. For PLD growth of BMT-PT films under the conditions investigated here, it was found that all films were nonstoichiometric. In particular, all films were Pb-deficient. Based on the composition data in Figs. 4 and 5, it is clear that a mixture of flux and adsorption control exists in this system. The B-site cations (Mg and Ti) appear to be flux controlled, as adding an excess of Mg leads to an increase in the Mg concentration. However, both A-site cations are adsorption controlled, where Bi adsorption controls the adsorption of Pb into the film. Pb and Bi have been shown to be adsorption controlled in PbTiO₃, Bi₄Ti₃O₁₂, and other films grown by molecular beam epitaxy [62]–[65].

Fig. 4 demonstrates that Pb is adsorption controlled, as changing the flux of Pb leads to very little overall change in the overall Pb concentration. All films were substoichiometric in Pb. Moreover, the Pb concentration in the film shows far less change than that of Bi with changes in target composition. It is hypothesized that preferential adsorption of Bi in the growing film leads to a fixed concentration of Pb-vacancies, in order to achieve electroneutrality. Considering the chemical formula as $(Bi_{0.35}Pb_{0.65})(Mg_{0.175}Ti_{0.825})O_3$ and the observed defects, several defect reactions may be considered

$$75PbO + 21MgO + 99TiO_{2} + 22Bi_{2}O_{3}$$

$$\xrightarrow{35BiMT-65PT} V_{Pb}'' + 2Bi_{Pb}^{\bullet} + 75Pb_{Pb}^{X} + 42Bi_{Bi}^{X} + 99Ti_{Ti}^{X} + 21Mg_{Mg}^{X} + 360O_{O}^{X}$$
(1)

$$\frac{\frac{1}{2}O_{2}(g) + 25PbO + 7MgO + 33TiO_{2} + 7Bi_{2}O_{3}}{\frac{35BiMT - 65PT}{\longrightarrow}V_{Pb}'' + 2h^{\bullet} + 25Pb_{Pb}^{X} + 14Bi_{Bi}^{X} + 33Ti_{Ti}^{X} + 7Mg_{Mg}^{X} + 120O_{O}^{X}}$$
(2)

and the electroneutrality condition is

$$[\operatorname{Bi}_{\operatorname{Pb}}^{\bullet}]2[V_O^{\bullet\bullet}] + p = 2[V_{\operatorname{Pb}}''] + 2[V_{\operatorname{Mg}}''] + n.$$
(5)

The lower volatility of Bi, with respect to Pb, is likely to favor excess Bi on Pb sites. Thus, to maintain electroneutrality, $[V_{Pb}'']$, $[V_{Mg}'']$, and/or the electron concentrations will increase. While the dominant defect reaction is not known, all of the films are reasonable insulators at low fields, suggesting that the compensation is predominantly ionic, or that any carriers are in deep traps. The observation of increased leakage currents at high fields does suggest that at least some of the compensation is electronic. In Fig. 3, as the [Mg] and, to a less extent, the [Pb] decrease, there is an increase in the Bi concentration, which is consistent with the defect reactions.

The complex defect chemistry of this system favors nonstoichiometry in films grown by PLD. A-site deficient films grown on PbTiO₃ seed layers to stabilize the perovskite phase showed P_r up to 22 μ C/cm² for films on Si. In contrast, bulk ceramics P_r is reported to be 33–38 μ C/cm² with similar compositions [15]. The decrease in the P_r with respect to bulk ceramics may be related to a combination of the tensile stress constraining the domain state, and the significant number of point defects in the grown films. More in depth electrical analysis on the electrical response of A-site deficient 35 BiMT - 65 PT thin films is discussed elsewhere [50].

V. CONCLUSION

 \approx 600–700-nm 35 BiMT-65 PT thin films were deposited using PLD. The film composition was studied as a function of the film deposition pressure, target composition, and laser deposition rate. The processing window for phase pure perovskite 35 BiMT - 65 PT was determined to be very wide in terms of deposition pressure, target composition, and laser deposition rate. EDS measurements showed that the [Mg] and [Pb] decreased with decreasing pressure, while the [Bi] increased. As the film target composition was adjusted, the [Bi] and [Mg] concentrations approached stoichiometric, but the [Pb] remained substoichiometric. Finally, the deposition rate showed no substantial effect on the film composition, but a significant impact on the ferroelectric properties. It was hypothesized that the preferential adsorption of Bi dictates the adsorption of Pb into the A-site. The $2P_r$ increases 64% as the deposition pressure increased and the P-E loops rotated counterclockwise. The P_r increases as the target Pb excess increases to 48 mol%. As the target Pb excess increased over 48 mol%, the P_r decreases possibly due to the need to evolve more PbO from defective growth layers. As the deposition rate decreased, the P_r increased due to enhanced crystalline quality. At laser frequencies of 5 Hz and below, a Mg-rich pyrochlore phase begins to form. A-site deficient 35 BiMT - 65 PT thin films showed a maximum $P_r \approx 22 \ \mu \text{C/cm}^2$.

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