

First-Principles Density Functional Theory Investigation of Oxygen and Water Adsorption on the (010), (100), and (111) Surfaces of BaZrS₃

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In the last decade, perovskites have attracted remarkable attention as promising alternatives to silicon in the solar industry to achieve more efficient and cost-effective solar cells owing to their outstanding optical and electronic properties and low synthesis costs. However, the conventional hybrid organic–inorganic halide perovskites are highly unstable and toxic, which has hindered their commercialization. Recently, chalcogenide perovskites (CPs) have emerged as promising candidates to overcome these challenges. Among CPs, BaZrS₃ has attracted more attention owing to its excellent optoelectronic properties and high stability. However, to guarantee long-term performance under atmospheric conditions, it is important to gain a fundamental understanding of the interactions between BaZrS₃ surfaces and atmospheric agents, such as O₂ and H₂O. Herein, we employed density functional theory to study the adsorption of H₂O and O₂ molecules on the (010), (100), and (111) surfaces of BaZrS₃. On all three surfaces, the Zr sites were more reactive than the Ba sites towards H₂O and O₂. Generally, O₂ molecules adsorbed more strongly at the Zr sites than H₂O molecules, with significant charge transfer from the interacting surface species. Consistent with their stronger adsorption, O₂ molecules drew more charges from the interacting species than H₂O, oxidizing them to a greater extent. With the surface structures preserved upon H₂O and O₂ adsorption, we observed only small adsorption-induced changes in electronic properties, with the covered surface exhibiting slightly smaller band gaps than the bare surfaces. These results demonstrate a high degree of thermodynamic and atmospheric stability of BaZrS₃ surfaces and show that, unlike halide perovskites, the exposure of BaZrS₃ thin-film surfaces to H₂O and O₂ molecules may not have significant detrimental effects on their PV performance.