

# Electrocaloric Effect of Perovskite High Entropy Oxide Films

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This paper describes two perovskite high entropy oxide (PHEO) compositions:  $\text{Pb}(\text{Hf}_{0.2}\text{Zr}_{0.2}\text{Ti}_{0.2}\text{Nb}_{0.2}\text{Mn}_{0.2})\text{O}_3$  (Mn PHEO) and  $\text{Pb}(\text{Hf}_{0.2}\text{Zr}_{0.2}\text{Ti}_{0.2}\text{Nb}_{0.2}\text{Al}_{0.2})\text{O}_3$  (Al PHEO). Powders are prepared by conventional solid state sintering by first pre-reacting the B-site oxides, then adding PbO. Phase pure Mn PHEO powder is obtained following calcination of the mixed powders at 750 ÅC for 240 min; however, secondary phases persisted in Al PHEO for heat treatments from 750 ÅC to 1200 ÅC. The Mn PHEO undergoes an entropy-driven phase transformation. Thin films of these compounds are synthesized by pulsed laser deposition (PLD) on a lead zirconate titanate seed layer on Pt-coated  $\text{SiO}_2/\text{Si}$ . The dielectric response of the Mn PHEO films show some contribution from space charge polarizability; in contrast, the Al PHEO films show a slim ferroelectric hysteresis loop and relaxor-like characteristics. The Al PHEO has a dielectric permittivity of  $\approx 2000$  with a loss tangent  $< 0.05$  from 100 Hz to 100 kHz; it has a dielectric maximum at 105 Å} 0.5 ÅC and a Burns' temperature of 234 Å} 0.5 ÅC. Indirect measurements based on the Maxwell-relations yielded a maximum electrocaloric temperature change of 8.4 K at 180 ÅC under the applied electric field of 1186 kV cm<sup>-1</sup>.